Geology, Mineral Deposits, and Geochemical and Radiometric Anomalies, Serpentine Hot Springs Area, Seward Peninsula, Alaska

By C. L. Sainsbury, Travis Hudson, Reuben Kachadoorian, and Thomas Richards

CONTRIBUTIONS TO ECONOMIC GEOLOGY

GEOLOGICAL SURVEY BULLETIN 1312-H

A description of mineralized faults and veins near the granite stock of Serpentine Hot Springs and the geochemical and radiometric methods used to find them
CONTENTS

Abstract ........................................................................................................... H1
Introduction ..................................................................................................... 1
Acknowledgments and analytical methods ...................................................... 2
Areal geology .................................................................................................. 4
  Metamorphic rocks ..................................................................................... 5
  Carbonate rocks ......................................................................................... 6
  Igneous rocks ............................................................................................. 6
Structure ......................................................................................................... 7
Mineralized areas .......................................................................................... 8
Radiometric anomalies .................................................................................. 10
Reconnaissance geochemical surveys ............................................................ 11
Detailed geochemical survey of Humboldt Creek ........................................... 12
Suggestions for exploration ........................................................................... 18
References cited ............................................................................................. 18

ILLUSTRATIONS

PLATE 1. Geologic map of the Serpentine Hot Springs area, Seward Pen-
insula, Alaska, showing location of bedrock and stream-sedi-
ment samples, and radiometric anomalies ................................................... In pocket

FIGURE 1. Map of Seward Peninsula, showing location of Serpentine-
Kougarok area .............................................................................................. H2
  2. Maps showing distribution of metals in panned concentrates
     and stream sediments, Humboldt Creek .................................................. 16

TABLES

TABLES 1–4. Metal contents of rock samples in Serpentine
     Hot Springs area:
     1. Unaltered rock units ........................................................................... H9
     2. Bedrock samples and panned concentrates from
        bedrock samples ................................................................................ In pocket
     3. Stream sediments .............................................................................. 13
     4. Stream sediments and panned concentrates of
        stream sediments, Humboldt Creek drainage ................................ In pocket
CONTRIBUTIONS TO ECONOMIC GEOLOGY

GEOLOGY, MINERAL DEPOSITS, AND GEOCHEMICAL AND RADIOMETRIC ANOMALIES, SERPENTINE HOT SPRINGS AREA, SEWARD PENINSULA, ALASKA

By C. L. SAINSBURY, TRAVIS HUDSON, REUBEN KACHADOORIAN, and THOMAS RICHARDS

ABSTRACT

Geologic mapping, analyses of samples of bedrock, and geochemical studies have disclosed the probable source of placer gold and tin on Humboldt Creek, Serpentine-Kougarok area, Alaska, and have shown mineralized bedrock in several areas on the east side of the granite stock at Serpentine Hot Springs. Gold, silver, mercury, arsenic, cobalt, copper, molybdenum, nickel, lead, antimony, tin, tungsten, and zinc occur in panned stream concentrates and in altered bedrock that is associated with steep faults, thrust faults, and quartz veins. Two mineralized and altered fault zones were sampled in detail; one contains float fragments of highly argentiferous galena (as much as 5,000 parts per million Ag). Anomalous amounts of metal in bedrock samples, in stream sediments, and in panned concentrates from stream sediments and rock samples suggest areas of mineralized bedrock outside the granite; radiometric anomalies lie within the granite and near the contact of the granite but have not been evaluated on the ground.

INTRODUCTION

In U.S. Geological Survey Circular 565 (Sainsbury and others, 1968), the large size and great quantity of placer cassiterite in Humboldt Creek on Seward Peninsula (fig. 1) were emphasized, and possible mineralized structures in bedrock southeast of the granite were discussed. During 1968, detailed mapping of Humboldt Creek and adjacent areas in the Serpentine-Kougarok area (pl. 1) was carried out as part of a program that included detailed geochemical studies and airborne geophysical surveys. Because a preliminary geochemical survey, using the -80-mesh fraction of stream sediments, failed to indicate the known placer cassiterite on Humboldt Creek, this creek...
was used as a test area to compare the data obtained from stream sediments and the data from panned concentrates. All work discussed herein is part of the Heavy Metals program of the U.S. Geological Survey.

ACKNOWLEDGMENTS AND ANALYTICAL METHODS

Analytical results are an essential starting point for the interpretive part of this study. Analysts are specifically credited on individual tables in this report, but we particularly acknowledge here the efforts of D. J. Grimes, R. L. Miller, K. J. Curry, R. B. Tripp, and W. R. Vaughn.

Fieldwork was facilitated by the help of numerous individuals. We are especially indebted to Maurice Kelleher, James Isbell, Mr. and Mrs. Robert Emmons of the Alaska Department of Highways, and Mr. Carl Melton of the Federal Aviation Agency. All samples were prepared for analyses by Clifford Weyiouanna and Duane Barnard, geologic field assistants.
Bedrock samples were crushed to -1/2-inch mesh in a jaw crusher, a split was saved, and the remainder was pulverized in a ceramic-plate pulverizer. A weighed portion of pulverized material was panned, and the panned concentrate was weighed and analyzed as a check against analyses of unpanned mineralized samples. Stream sediments were collected from behind rocks on stream bars, then air dried and screened through a 40-mesh plastic screen; each of the total samples was later pulverized to -200 mesh. Pan concentrates from Humboldt Creek were air dried, a weighed portion was retained for mineralogical studies, and the remainder was pulverized. After pulverizing each sample, the pulverizer plates were cleaned by pulverizing a teaspoonful of white quartz, and the resulting powder was added to the sample previously pulverized. This was done to prevent contamination of later samples, because laboratory experiments by J. C. Antweiler (oral commun., 1967) have shown that even small gold particles may smear onto the pulverizer plates and register in the following sample.

Samples were analyzed for gold by atomic absorption, for mercury by mercury detector, and for other elements by semiquantitative spectrographic analyses. Some duplicate splits of samples were analyzed for copper, lead, zinc, tin, and arsenic by wet chemical methods. With the exception of results from gold analyses, which were erratic, analytical results generally were in good agreement.

Only the salient features of the analytical data are presented on the maps that accompany this report. The individual analyses are presented as tables 2-4.

Analytical data were selected and plotted on plate 1 and in figure 2 according to the following method:

1. Background values of elements in various lithologic units were selected according to results of analyses of samples of unaltered rock units (table 1).

2. Only elements that are anomalously high in the selected specimens of argentiferous galena, or in samples of altered rock along faults and veins, are considered. These elements include gold, silver, mercury, arsenic, cobalt, copper, molybdenum, nickel, lead, antimony, tin, tungsten, and zinc.

3. Elements present in each sample in amounts above background values were noted, and a numerical value of the anomaly for each metal was determined by dividing the total content for that metal by the background value shown in table 1. This gives a ratio in which the background is represented by the number 1. If the numbers representing the concentration ratios are added and the sum of the backgrounds is subtracted, this gives a figure which represents the magnitude of the total anomaly at that
sample site. For instance, a sample that contains 15 ppm (parts per million) Mo, 100 ppm Sn, and 15 ppm Ag would be treated as follows: 15 + 5 = 3 (the concentration ratio), 100 + 15 = 6.6, 15 + 1 = 15, and 3 + 6.6 + 15 = 24.6 (the total concentration ratio). The anomaly, however, is 24.6 - 3 (the sum of the three backgrounds represented by the number 1 for each element present in more than background concentration), or 21.6.

4. Symbols are shaded to show the number of metals present in anomalous amount in that sample; the size of the shaded symbol is varied to show the total anomaly. The map symbol shows at a glance the salient features important in geochemical reconnaissance surveys—that is, the number of anomalous metals in the sample and the magnitude of the anomaly. The detailed analytical data are presented in tables 2-4.

**AREAL GEOLOGY**

The bedrock geology of the area is shown on plate 1, which synthesizes data obtained by all authors of this report. Except for the granite, the distribution of the major lithologic units is controlled mainly by thrust faulting. The main lithologic units consist of (1) the granite stock, which is a porphyritic biotite granite, (2) intensely deformed and sheared metamorphic rocks of Precambrian age, and (3) younger carbonate rocks regionally metamorphosed to marbles with some relict beds and areas of limestone and dolomite. The older carbonate rocks are equivalent to rocks of pre-Ordovician age mapped in the York Mountains, about 80 miles west. In the York Mountains these rocks consist of thinly bedded dolomitic limestone and argillaceous limestone (Sainsbury, 1965); in the area discussed in this report, they are moderately metamorphosed to the stage in which the argillaceous bands are noticeably micaceous. The younger carbonate rocks, which originally consisted of medium- to thick-bedded light-gray to dark-gray limestone locally converted to dolomite along thrusts and normal faults, are now largely sugary-textured marbles in which original bedding is still discernible. These younger marbles are confined to thrust plates and are correlative with Paleozoic carbonate rock exposed discontinuously in the thrust plates of the Collier thrust belt (Sainsbury, 1969), which covers the entire Seward Peninsula. Fossils of probable Devonian age have been recovered from marbles in thrust sheets about 15 miles south of the area of this report, whereas fossils of Ordovician, Silurian, and Devonian age have been recovered from unmetamorphosed limestone in the York Mountains.

---

1 Travis Hudson is continuing a detailed study of the granite and the rocks surrounding it on the east, which will result in some modification of the geology shown on plate 1.
Two main units of high-rank metamorphic rocks (gneiss and chloritic schist) and one thick unit of moderately metamorphosed dark graphitic siltite, with numerous minor variants such as graphitic slate, schist, marble, and graywacke, are shown on plate 1. Probably the oldest rock is the orthogneiss which is exposed over several square miles northwest of the granite. The gneiss, which is gray to pinkish gray where fresh, is composed of quartz, pink orthoclase, and minor amounts of biotite and superficially resembles the granite. Southward, the gneiss grades into light-colored leucocratic schist and gneiss with residual marble beds largely converted to calc-silicate rock. The schist and gneiss is intruded and cut off by the granite.

The next oldest high-rank metamorphic rock unit, confined to the east side of the granite, consists of chlorite-epidote-amphibole schists with intercalated schistose marble and metamorphosed mafic intrusive rocks (metagabbro of table 1). Glauconite and blue-green amphibole are common; these schists are retrograded blueschist facies rocks (Sainsbury and others, 1970), which crop out discontinuously over hundreds of square miles to the south and southwest and which probably are exposed in thrust slices as far as the glauconite-bearing rocks described by Smith (1910) in the Solomon and Casa-depaga quadrangles east of Nome. These rocks form part of the Nome Group, originally defined as of Paleozoic age (Brooks and others, 1901) on the basis of fossils collected from limestones within the chloritic schists. Mapping by Sainsbury from 1960 to 1967 has demonstrated that the fossiliferous limestones, formerly thought to be intercalated in these chloritic schists of the Nome Group, are thrust slices of younger (Paleozoic) limestones. The chloritic schists, which represent highly metamorphosed and deformed rocks that originally contained abundant iron and magnesium, are thought by Sainsbury to be of Precambrian age. The chloritic schists everywhere observed are intensely deformed and folded; in the area of this report, recumbent small-scale isoclinal folds are overturned to the west.

The moderately metamorphosed rock unit shown as metasiltite and related rocks on plate 1 exhibits several variants which are not differentiated in this report. These variants reflect original variations in lithology and changes in mineralogy due to regional metamorphism, intense shearing during thrusting, and thermal metamorphism. The original rock was a graphitic siltite composed of silt-sized quartz grains and abundant carbonaceous material, and a variable content of calcite. Thin beds of carbonaceous limestone, shale, graywacke, and quartz arenite are now represented by graphitic marble schist, slate, and cleaved rocks with abundant veinlets of white to vitreous
quartz. In the thermal aureole around the granite, slate has gone to
hornfels and biotite-tourmaline rock; calcareous units are converted
to calc-silicate rock. All the described variants are included in the
map unit (Kuzitrin Series of Brooks and others, 1902), although
detailed mapping by Hudson on the east side of the granite has
demonstrated that individual beds can be mapped. In this area,
the graphitic rocks are intensely deformed, with small-scale isoclinal
folds overturned to the west; the degree of deformation increases
near thrust faults, as does the number of vitreous quartz veinlets.

The graphitic rocks are clearly of pre-Ordovician age in the York
Mountains (Collier, 1902; Sainsbury, 1965), because they underlie
thin-bedded argillaceous limestones, equivalent to those of this
report. Because they can be traced eastward to merge with the
graphitic rocks of the Serpentine-Kougarok area, all are considered
to be of pre-Ordovician age. Sainsbury believes that they are
probably of late Precambrian age, following the reasoning first
advanced by Collier (1902, p. 17).

CARBONATE ROCKS

Two distinctly different types of carbonate rocks crop out east of
the granite; both are in thrust-fault contact with older metamorphic
rocks. Rocks of the older carbonate unit were described on page H5.
The younger carbonate rocks, confined to thrust plates east of
Humboldt Creek (pl. 1), are moderately folded but are not schistose.
Because of regional thermal metamorphism, they were largely con-
verted to marble in which, however, original bedding can be seen.
As is common in the York Mountains, about 80 miles west, the
limestones were dolomitized near thrusts and normal faults prior to
regional metamorphism. In the area of this report, the marbles
near thrust faults locally have been replaced by silica, and the under-
laying rocks have been altered and stained. In the small klippe
southeast of Humboldt Creek, xenoliths of slate are abundant locally
and represent fragments of underlying rocks carried up along thrust
slices.

IGNEOUS ROCKS

The igneous rocks that crop out in the area of plate 1 consist of the
porphyritic biotite granite stock, the related small fine-grained
granitic dikes that occupy joints in the granite and the surrounding
rocks, and a few dark dikes of diabase and lamprophyre. None of the
dikes have been traced in detail. The lamprophyres near the granite
contain corroded xenocrysts of quartz and orthoclase, a character-
istic of lamprophyre dikes near granite of the western Seward Pe-
ninsula (Sainsbury, 1969). Farther away, the dark dikes are typical
diabase, unsheared and unaltered, which shows that they were injected after the deformation and thermal metamorphism of the enclosing rocks.

The granite was described by Moxham and West (1953). It consists of unfoliated biotite granite with orthoclase crystals as much as 1 inch long. A distinct border facies is marked by a color change from pink orthoclase outward to white orthoclase and by a slight decrease in grain size. The border facies is mapped only on the northwestern and southwestern boundaries of the granite, where a large fault zone is inferred largely on the basis of the truncated border facies. Other, more subtle variants are being mapped and sampled in detail by Travis Hudson.

The absolute age of the granite at Serpentine Hot Springs is unknown. Granitic plutons in the eastern Seward Peninsula have been assigned a Late Cretaceous age on the basis of potassium-argon age determinations (Miller and others, 1966). The granite at Brooks Mountain, in the York Mountains, has been dated as Late Cretaceous (Sainsbury, 1969), and the associated diabase and lamprophyre are younger and are considered to be of Late Cretaceous to early Tertiary age. At Serpentine Hot Springs, as in the York Mountains, ore deposition is younger than the injection of both the granite and the dike rocks.

STRUCTURE

The mapped area lies within the Collier thrust belt, and the distribution and structure of rock units older than the granite are controlled in large part by thrusts. Just east of the area of plate 1, effects of thrusting are so complex that mapping at a scale larger than mile-to-the-inch will be required to decipher the structure. On the basis of existing mapping, two alternate interpretations are possible of the major structure in the area of plate 1. The first, supported by dips in the graphitic slate on the west side of the area, is that the oldest metamorphic rock (gneiss) is exposed along the axis of a sharp fold that trends north and that younger slate is exposed on the west flank. The second interpretation, supported by the known thrusts and by the small-scale isoclinal folds overturned to the west, is that the slate is thrust over gneiss and over itself. Clearly, slate is thrust over slate off the southeastern part of the granite, and marble is in thrust-fault contact with older rocks.

After the thrusting, the granite was emplaced, and then the granite and the thrust plates were cut by several sets of steep faults. The faults conform generally to sets striking about north to N. 15° E., about N. 30° E., and about east—the youngest set strikes N. 15°–50° W. An intricate network of faults southeast of the granite is
associated with major geochemical anomalies, and the main mineralized areas lie along altered faults that strike about N. 50° W.

MINERALIZED AREAS

Two main areas of mineralized bedrock were sampled in detail, after initial samples collected by Hudson in the area presumed to be favorable on the basis of work done in 1967 by C. L. Sainsbury, Reuben Kachadoorian, T. E. Smith, and W. C. Todd were found to be highly anomalous in metals. As here defined, "anomalous" values are those that exceed the maximum content found in unaltered rock units, as shown in table 1. One area represented by bedrock samples 56–61 (pl. 1) consists of an altered zone that strikes about N. 55° W. across a saddle southwest of the south headwaters of Humboldt Creek. Float of rusty fracture fillings and rusty quartz lies along the zone, which can be traced at least 2,000 feet. Bulk samples of float quartz and altered graphitic siltite along the zone contain anomalous amounts of many metals (table 2), and a grab sample of rusty float contains highly anomalous amounts of gold, silver, mercury, arsenic, copper, molybdenum, lead, antimony, and zinc, amounting to more than 1,000 times the total background value of these metals. These samples were collected over a width of 200 feet and a length of 1,000 feet along the flat saddle, where frost action has completely shattered bedrock to create a veneer of surface rubble. Hence, nothing can be stated as to the width of possible veins that exist within the altered zone beneath the frost-shattered rock. Samples that contained only a few metals in anomalous amount yielded panned concentrates that contained, in addition, many related metals in anomalous amount. The above suite of associated metals is that commonly found in the silver zone of tin deposits (Sainsbury and Hamilton, 1968, p. 329, 330).

In a second area, silver-rich galena crops out on the south side of the southwestern headwaters of Humboldt Creek. Here an altered and stained fault zone can be traced for at least 2,500 feet, and it is probably continuous for an additional 2,000 feet. Numerous samples, represented by sample localities 25, 26, and 55 (pl. 1), contain highly anomalous amounts of gold, silver, lead, mercury, arsenic, molybdenum, antimony, tin, copper, and tungsten, all of which are enriched in the hand specimens of argentiferous galena. Float fragments of galena occur only in a small area on the slope break of the drainage, in what could be an old prospect pit that is almost completely obliterated by creep. Samples of frost-riven float of altered graphitic slate and stained quartz taken over an area of 1,000 feet by 200 feet along the altered fault contained highly anomalous amounts of metal. Again, panned concentrates showed a great increase in number of anomalous metals. A sample of stained quartz
TABLE 1.—Metal content, in parts per million, of unaltered rock units, Serpentine Hot Springs area

(All analyses are semiquantitative spectrographic, except those for mercury, which are mercury detector. Semiquantitative spectrographic analyses are reported in parts per million to the nearest number in the series 0.5, 0.7, 1.0, 1.5, 3.0, 5.0, 7.0, 10, 15, which represent points on a geometric scale. The precision of a reported value is approximately ±100 percent or ±50 percent. Semiquantitative spectrographic analyses determined by J. C. Hamilton and K. J. Curry; mercury detector analyses, by R. L. Miller and W. R. Vaughn.)

<table>
<thead>
<tr>
<th>Rock unit</th>
<th>Hg</th>
<th>Mn</th>
<th>Ba</th>
<th>Co</th>
<th>Cu</th>
<th>Mo</th>
<th>Ni</th>
<th>Pb</th>
<th>Sn</th>
<th>Zn</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>1</td>
<td>0.04</td>
<td>150</td>
<td>150</td>
<td>&lt;5</td>
<td>5</td>
<td>5</td>
<td>70</td>
<td>15</td>
<td>&lt;200</td>
</tr>
<tr>
<td>Granitic dike</td>
<td>2</td>
<td>0.04-0.09</td>
<td>200-400</td>
<td>500</td>
<td>6-7</td>
<td>15-26</td>
<td>&lt;5-7</td>
<td>70</td>
<td>10</td>
<td>&lt;200</td>
</tr>
<tr>
<td>Slate (graphitic schist map unit)</td>
<td>3</td>
<td>0.01-0.03</td>
<td>1,000-1,500</td>
<td>300-1,000</td>
<td>15</td>
<td>7-20</td>
<td>&lt;5-7</td>
<td>50-100</td>
<td>10-30</td>
<td>&lt;10</td>
</tr>
<tr>
<td>Chloritic schist and metagabbro</td>
<td>3</td>
<td>0.01-0.02</td>
<td>300-700</td>
<td>200-300</td>
<td>5-20</td>
<td>5-100</td>
<td>&lt;5</td>
<td>20-60</td>
<td>&lt;5-10</td>
<td>&lt;10</td>
</tr>
</tbody>
</table>
collected 2,000 feet away along a narrow altered zone that is the probable continuation of the mineralized fault contained 15 ppm Ag, and anomalous gold, copper, and molybdenum (sample loc. 55, pl. 1). Continuity between these sample localities is assumed only, because talus mantles the slopes between.

Numerous other samples of altered zones contain anomalous amounts of several metals, especially in the highly faulted area on the southeastern side of the granite. The limited sampling suggests that faults that trend slightly east of north commonly contain quartz vein material that is auriferous but of low grade, whereas faults that trend about N. 50° W. are more highly mineralized with argentiferous galena.

Several samples of altered and silicified marble collected above the thrusts east of Humboldt Creek show anomalous amounts of several metals, as did samples of altered schist collected between the klippe of marble. No attempt has been made to outline the extent of the altered zones or the concentration of metal along these zones. Elsewhere in the Serpentine-Kougarok area, where silicified zones along thrust have been sampled, only minor amounts of copper, with traces of gold, were introduced with the silica. None of the copper-bearing silicified replacement zones elsewhere are closely associated with granite or with mineralized faults; hence, the authors attach more importance to the altered rocks adjacent to the thrust faults near Humboldt Creek because many different metals characteristic of the altered and mineralized rocks near the granite are found in altered rocks along the thrust faults near Humboldt Creek.

RADIOMETRIC ANOMALIES

Moxham and West (1953) showed that the granite contains small amounts of radioactive material disseminated throughout, principally as radioactive primary accessory minerals such as zircon, sphene, and allanite; hydrogoethite and two other, unidentified, secondary minerals also are radioactive. The radioactivity in the primary minerals is caused largely by thorium, whereas that in the secondary minerals is caused by uranium. Placer concentrates, obtained by Moxham and West from the surface of gravel bars, contained zircon, sphene, and allanite as the principal radioactive minerals. Moxham and West concluded (p. 8) that “The investigations in the Serpentine-Kougarok area failed to reveal either placer or bedrock deposits of radioactive material in quantities of present commercial interest.”

An airborne magnetic and radiometric survey was made for the U.S. Geological Survey by Lockwood, Kessler and Bartlett, Inc. after completion of the fieldwork. Several radiometric anomalies were
SERPENTINE HOT SPRINGS AREA, ALASKA

outlined and are reproduced on plate 1. None have been field checked. However, the following factors are significant:

1. Although several broad anomalies within the granite tend to correspond to topographic highs, the area with the highest radiation lies along the southeastern edge of the granite in an area not checked in detail by Moxham and West. The single uranium deposit known on the Seward Peninsula, at Brooks Mountain, about 80 miles west, consists of secondary uranium minerals that are associated with a lobe of granite (West and White, 1952). The similarity of geology and location of radiometric anomalies may be significant. Work by Travis Hudson in 1969 shows that the granite in this area may be a late-stage differentiate.

2. The radiometric anomaly does not coincide with known mineralized bedrock; hence, no direct correlation can be made between anomalies and ore deposits.

3. All the anomalies occur above the granite or the orthogneiss, which suggests that radioactive accessory minerals may be the principal source of the radiation.

Factors 2 and 3 tend to lessen the possible importance of factor 1.

RECONNAISSANCE GEOCHEMICAL SURVEYS

A reconnaissance geochemical survey in 1967, using only the -80-mesh fraction of stream sediments, did not show anomalous tin on Humboldt Creek, where cassiterite in nuggets as much as 3 inches in diameter was observed in sluice-box concentrates from placer gold mining. Consequently, alternative methods were applied in 1968 to determine the applicability of different geochemical techniques in this area, where heavy and chemically resistant minerals (gold and cassiterite) were expected to be found in association with base-metal sulfide minerals. These methods included collection and analyses of the -40-mesh fraction of stream sediments collected from gravel bars or in the lee of boulders, where resistant heavy minerals accumulate to some degree, of panned concentrates obtained from stream gravels, and of sluice-box concentrates obtained by passing about a yard of gravel through a portable sluice box.

The geochemical data, plotted according to the same method used for bedrock data, are summarized on plate 1 and in table 3. In general, both -80-mesh and -40-mesh fractions of stream sediments show only relatively low anomalies with respect to the background values of rock units in the drainage basins of the streams sampled. Anomalies in sediments must be interpreted on the basis of one to three metals in low amounts—generally much lower than five times
background. This is in striking contrast to results obtained from the
-40-mesh fraction of stream sediments collected near mineralized
bedrock in the York Mountains, where highly anomalous amounts
of tin, beryllium, copper, lead, zinc, arsenic, antimony, and tungsten
were found (Sainsbury, Hamilton, and Huffman, 1968). Background
values for tin in the York Mountains averaged less than 20 ppm;
near lodes, stream sediments contained as much as 1,100 ppm. Hence,
the failure of stream sediments to point to the lodes found subse-
quently in the drainages of Humboldt Creek, by sampling and analyses
of altered bedrock, is noteworthy. The exploration geologist in search
of mineral deposits in this part of Alaska, especially placer deposits,
would do well to apply several of the known methods of geochemical
exploration, especially panning.

DETAILED GEOCHEMICAL SURVEY OF HUMBOLDT CREEK

The results of detailed studies along Humboldt Creek are summa-
rized in figure 2 and in table 4. To supplement the stream-sediment
survey, numerous samples of concentrates were collected by panning
stream gravels as well as alluvium in cutbanks. Concentrates were air
dried and weighed, a split was saved for mineralogical work, and the
remainder of the concentrates was pulverized and analyzed by the
same methods as those used for bedrock and stream-sediment samples
(p. H3). The results are shown in figure 2. Part A of figure 2 includes
the gold analyses; part B has gold excluded, for the simple reason that
the magnitude of the total anomaly at many sample sites is controlled
largely by gold. Exclusion of gold, which occurs in a highly anomalous
amount in many panned concentrates, allows a more dependable
comparison with the other geochemical results.

Figure 2 depicts several interesting facts. Tin was found in panned
concentrates of surface stream gravels only in the east fork of Hum-
boldt Creek, where several samples contain the metal in anomalous
amounts. In the sluice-box concentrate (loc. 15, fig. 2; table 4),
tin and several other metals were detected in highly anomalous
amounts. The metals that were concentrated in the argentiferous
galena (gold, silver, lead, arsenic, cobalt, copper, molybdenum,
nickel, antimony, tungsten, tin, zinc, and mercury) are com-
monly associated only in the panned concentrate. If the total con-
centration of elements, in parts per million, is divided by the
concentration ratio (weight of total sample divided by weight of con-
centrate), most of the anomalies in concentrates disappear; never-
theless, analyses of panned concentrates would lead one directly
to the base of the known outcrop of galena from a point far down-
stream. Analyses of samples of stream sediments, however, would
fail to do so, unless the samples were collected very near the lodes.
TABLE 3.—*Metal content, in parts per million, of stream sediments, Serpentine Hot Springs area, Seward Peninsula, Alaska*  

[N, below detection limit; L, detected but below limit of dependable reading; ND, not determined; ...., no anomaly. All sample numbers with laboratory numbers preceded by ACB or ACM represent 80-mesh fraction of total stream sediments; all preceded by AH represent 40-mesh fraction of total stream sediments. All analyses are semiquantitative spectrographic except those for Au, which are atomic absorption; those for Hg, which are mercury detector; and those in italic for Ag, Co, Ni, Pb, Sn, and Zn, which are wet chemical. Analysts: K. J. Curry, R. L. Miller, R. B. Tripp, and W. R. Vaughn. In backgrounds, numbers beneath element symbols give background values and, in parentheses, N values].

<table>
<thead>
<tr>
<th>Locality No. (pl. 1)</th>
<th>Laboratory No.</th>
<th>Field No.</th>
<th>Au (&lt;0.02)</th>
<th>Ag (&lt;0.05)</th>
<th>As (&lt;0.05)</th>
<th>Cu (&lt;5)</th>
<th>Mo (&lt;5)</th>
<th>Ni 150</th>
<th>Pb 70</th>
<th>Sh L (&lt;150)</th>
<th>Sn 15</th>
<th>W L (&lt;200)</th>
<th>Zn L (&lt;200)</th>
<th>Metals present</th>
<th>Total anomaly</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>ACM-531</td>
<td>07-AK-531</td>
<td>&lt;0.02</td>
<td>0.04</td>
<td>&lt;10</td>
<td>15</td>
<td>15</td>
<td>30</td>
<td>N</td>
<td>88-30</td>
<td>&lt;85.15</td>
<td>N</td>
<td>N</td>
<td>66, L</td>
<td>3</td>
</tr>
<tr>
<td>2</td>
<td>523</td>
<td>533</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>7</td>
<td>5</td>
<td>N</td>
<td>16</td>
<td>23</td>
<td>N</td>
<td>&lt;85.15</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>3</td>
<td>ACM-531</td>
<td>07-AK-531</td>
<td>&lt;0.02</td>
<td>N</td>
<td>0.03</td>
<td>L</td>
<td>15</td>
<td>15</td>
<td>N</td>
<td>80</td>
<td>10</td>
<td>N</td>
<td>L</td>
<td>N</td>
<td>10</td>
</tr>
<tr>
<td>4</td>
<td>523</td>
<td>533</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>3</td>
<td>3</td>
<td>N</td>
<td>80</td>
<td>30</td>
<td>N</td>
<td>80,500</td>
<td>L</td>
<td>N</td>
</tr>
<tr>
<td>5</td>
<td>523</td>
<td>533</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>1.5</td>
<td>1.5</td>
<td>N</td>
<td>3</td>
<td>30</td>
<td>N</td>
<td>80,150</td>
<td>N</td>
<td>N</td>
</tr>
<tr>
<td>6</td>
<td>523</td>
<td>533</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>7</td>
<td>15</td>
<td>N</td>
<td>7</td>
<td>70</td>
<td>N</td>
<td>&lt;85.15</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>7</td>
<td>533</td>
<td>533</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>5</td>
<td>15</td>
<td>N</td>
<td>15</td>
<td>30</td>
<td>N</td>
<td>&lt;85.15</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>8</td>
<td>533</td>
<td>533</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>7</td>
<td>15</td>
<td>N</td>
<td>15</td>
<td>30</td>
<td>N</td>
<td>&lt;85.15</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>9</td>
<td>544</td>
<td>544</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>15</td>
<td>15</td>
<td>3</td>
<td>30</td>
<td>30</td>
<td>N</td>
<td>&lt;85.15</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>10</td>
<td>544</td>
<td>544</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>10</td>
<td>10</td>
<td>ND</td>
<td>8</td>
<td>80</td>
<td>ND</td>
<td>110</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>11</td>
<td>547</td>
<td>547</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>15</td>
<td>30</td>
<td>N</td>
<td>15</td>
<td>15</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>12</td>
<td>546</td>
<td>546</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>15</td>
<td>30</td>
<td>N</td>
<td>15</td>
<td>15</td>
<td>N</td>
<td>&lt;85.15</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>13</td>
<td>546</td>
<td>546</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>15</td>
<td>30</td>
<td>N</td>
<td>15</td>
<td>15</td>
<td>N</td>
<td>&lt;85.15</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>14</td>
<td>544</td>
<td>544</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>20</td>
<td>30</td>
<td>N</td>
<td>20</td>
<td>30</td>
<td>N</td>
<td>&lt;85.15</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>15</td>
<td>ACM-531</td>
<td>89-AK-531</td>
<td>&lt;0.02</td>
<td>N</td>
<td>0.06</td>
<td>N</td>
<td>50</td>
<td>50</td>
<td>N</td>
<td>50</td>
<td>50</td>
<td>N</td>
<td>50</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>16</td>
<td>541</td>
<td>541</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>15</td>
<td>20</td>
<td>3</td>
<td>30</td>
<td>30</td>
<td>N</td>
<td>&lt;85.15</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>17</td>
<td>541</td>
<td>541</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>15</td>
<td>20</td>
<td>3</td>
<td>30</td>
<td>30</td>
<td>N</td>
<td>&lt;85.15</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>18</td>
<td>ACM-531</td>
<td>89-AK-531</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>15</td>
<td>20</td>
<td>3</td>
<td>30</td>
<td>30</td>
<td>N</td>
<td>&lt;85.15</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>19</td>
<td>546</td>
<td>546</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>20</td>
<td>50</td>
<td>N</td>
<td>50</td>
<td>100</td>
<td>N</td>
<td>70</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>20</td>
<td>544</td>
<td>544</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>15</td>
<td>15</td>
<td>N</td>
<td>70</td>
<td>70</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>21</td>
<td>544</td>
<td>544</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>100</td>
<td>100</td>
<td>N</td>
<td>70</td>
<td>70</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>22</td>
<td>544</td>
<td>544</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>7</td>
<td>15</td>
<td>N</td>
<td>15</td>
<td>70</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>23</td>
<td>ACM-531</td>
<td>89-AK-531</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>100</td>
<td>100</td>
<td>N</td>
<td>100</td>
<td>100</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>24</td>
<td>ACM-531</td>
<td>89-AK-531</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>100</td>
<td>100</td>
<td>N</td>
<td>100</td>
<td>100</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td></td>
</tr>
<tr>
<td>25</td>
<td>546</td>
<td>546</td>
<td>&lt;0.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>15</td>
<td>15</td>
<td>N</td>
<td>70</td>
<td>70</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td></td>
</tr>
</tbody>
</table>

See footnote at end of table.
<table>
<thead>
<tr>
<th>Locality No.</th>
<th>Laboratory No.</th>
<th>Field No.</th>
<th>Au &lt;0.02</th>
<th>Ag (0.5)</th>
<th>Hg &lt;10</th>
<th>Ag L &lt;100</th>
<th>Cu 20</th>
<th>Mn 6</th>
<th>Ni 150</th>
<th>Pb 70</th>
<th>Sh L &lt;150</th>
<th>Sn L &lt;10</th>
<th>W L &lt;200</th>
<th>Zn L &lt;200</th>
<th>Metals present in anomalous amounts and their approximate concentration ratio</th>
<th>Total anomaly</th>
</tr>
</thead>
<tbody>
<tr>
<td>26</td>
<td>ACB-326</td>
<td>233</td>
<td>&lt;0.02</td>
<td>0.7</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>57</td>
<td>ND</td>
<td>50</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Hg, 1.25)</td>
<td>20</td>
</tr>
<tr>
<td>27</td>
<td>ACB-324</td>
<td>234</td>
<td>&lt;0.02</td>
<td>0.6</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Pb, 3.14)</td>
<td>10</td>
</tr>
<tr>
<td>28</td>
<td>AHI-026</td>
<td>235</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>24</td>
<td>ND</td>
<td>20</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Sn, 2.38)</td>
<td>10</td>
</tr>
<tr>
<td>29</td>
<td>ACB-326</td>
<td>236</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Zn, 2.57)</td>
<td>10</td>
</tr>
<tr>
<td>30</td>
<td>237</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Cu, 1.37)</td>
<td>10</td>
<td></td>
</tr>
<tr>
<td>31</td>
<td>237</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Cu, 1.37)</td>
<td>10</td>
<td></td>
</tr>
<tr>
<td>32</td>
<td>AHI-027</td>
<td>238</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Cu, 1.37)</td>
<td>10</td>
</tr>
<tr>
<td>33</td>
<td>239</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Cu, 1.37)</td>
<td>10</td>
<td></td>
</tr>
<tr>
<td>34</td>
<td>ACB-239</td>
<td>240</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Cu, 1.37)</td>
<td>10</td>
</tr>
<tr>
<td>35</td>
<td>AHI-028</td>
<td>241</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Cu, 1.37)</td>
<td>10</td>
</tr>
<tr>
<td>36</td>
<td>ACB-240</td>
<td>242</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Cu, 1.37)</td>
<td>10</td>
</tr>
<tr>
<td>37</td>
<td>AHI-029</td>
<td>243</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Cu, 1.37)</td>
<td>10</td>
</tr>
<tr>
<td>38</td>
<td>244</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Cu, 1.37)</td>
<td>10</td>
<td></td>
</tr>
<tr>
<td>39</td>
<td>AHI-030</td>
<td>245</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Cu, 1.37)</td>
<td>10</td>
</tr>
<tr>
<td>40</td>
<td>246</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Cu, 1.37)</td>
<td>10</td>
<td></td>
</tr>
<tr>
<td>41</td>
<td>247</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Cu, 1.37)</td>
<td>10</td>
<td></td>
</tr>
<tr>
<td>42</td>
<td>248</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Cu, 1.37)</td>
<td>10</td>
<td></td>
</tr>
<tr>
<td>43</td>
<td>ACH-017</td>
<td>249</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Cu, 1.37)</td>
<td>10</td>
</tr>
<tr>
<td>44</td>
<td>ACH-018</td>
<td>250</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Cu, 1.37)</td>
<td>10</td>
</tr>
<tr>
<td>45</td>
<td>251</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Cu, 1.37)</td>
<td>10</td>
<td></td>
</tr>
<tr>
<td>46</td>
<td>ACH-019</td>
<td>252</td>
<td>&lt;0.02</td>
<td>0.5</td>
<td>0.06</td>
<td>ND</td>
<td>ND</td>
<td>31</td>
<td>ND</td>
<td>30</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>(Cu, 1.37)</td>
<td>10</td>
</tr>
</tbody>
</table>

**Table 3.—Metal content, in parts per million, of stream sediments, Serpentine Hot Springs area, Seward Peninsula, Alaska—Continued**
Where duplicate values are reported, highest value was used to compute anomaly.

<table>
<thead>
<tr>
<th></th>
<th>AHI-038</th>
<th>65-A 51-27</th>
<th>&lt;.02</th>
<th>N</th>
<th>ND</th>
<th>N</th>
<th>8</th>
<th>1.5</th>
<th>N</th>
<th>3</th>
<th>30</th>
<th>N</th>
<th>15</th>
<th>N</th>
<th>N</th>
<th>N</th>
<th>N</th>
<th>N</th>
<th>N</th>
<th>N</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>AHI-044</td>
<td>66-A 51-28</td>
<td>&lt;.02</td>
<td>N</td>
<td>ND</td>
<td>L</td>
<td>N</td>
<td>N</td>
<td>7</td>
<td>3</td>
<td>N</td>
<td>7</td>
<td>12</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
</tr>
<tr>
<td></td>
<td>ACM-630</td>
<td>67-A 51-290</td>
<td>&lt;.02</td>
<td>N</td>
<td>ND</td>
<td>L</td>
<td>N</td>
<td>N</td>
<td>7</td>
<td>3</td>
<td>N</td>
<td>7</td>
<td>12</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
</tr>
<tr>
<td></td>
<td>ACM-639</td>
<td>67-A 51-292</td>
<td>&lt;.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>7</td>
<td>3</td>
<td>N</td>
<td>7</td>
<td>12</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
</tr>
<tr>
<td></td>
<td>ACM-777</td>
<td>68-A 51-2</td>
<td>&lt;.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>7</td>
<td>3</td>
<td>N</td>
<td>7</td>
<td>12</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
</tr>
<tr>
<td></td>
<td>ACM-813</td>
<td>67-A 51-413</td>
<td>&lt;.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>7</td>
<td>3</td>
<td>N</td>
<td>7</td>
<td>12</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
</tr>
<tr>
<td></td>
<td>ACM-817</td>
<td>67-A 51-417</td>
<td>&lt;.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>7</td>
<td>3</td>
<td>N</td>
<td>7</td>
<td>12</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
</tr>
<tr>
<td>70</td>
<td>ACM-613</td>
<td>67-A 51-613</td>
<td>&lt;.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>7</td>
<td>3</td>
<td>N</td>
<td>7</td>
<td>12</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
</tr>
<tr>
<td>71</td>
<td>ACM-617</td>
<td>67-A 51-617</td>
<td>&lt;.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>7</td>
<td>3</td>
<td>N</td>
<td>7</td>
<td>12</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
</tr>
<tr>
<td>72</td>
<td>ACM-817</td>
<td>67-A 51-817</td>
<td>&lt;.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>7</td>
<td>3</td>
<td>N</td>
<td>7</td>
<td>12</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
</tr>
<tr>
<td>73</td>
<td>ACM-811</td>
<td>67-A 51-811</td>
<td>&lt;.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>7</td>
<td>3</td>
<td>N</td>
<td>7</td>
<td>12</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
</tr>
<tr>
<td>74</td>
<td>ACM-411</td>
<td>67-A 51-411</td>
<td>&lt;.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>7</td>
<td>3</td>
<td>N</td>
<td>7</td>
<td>12</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
</tr>
<tr>
<td>75</td>
<td>ACM-413</td>
<td>67-A 51-413</td>
<td>&lt;.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>7</td>
<td>3</td>
<td>N</td>
<td>7</td>
<td>12</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
</tr>
<tr>
<td>76</td>
<td>ACM-513</td>
<td>67-A 51-513</td>
<td>&lt;.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>7</td>
<td>3</td>
<td>N</td>
<td>7</td>
<td>12</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
</tr>
<tr>
<td>77</td>
<td>ACM-517</td>
<td>67-A 51-517</td>
<td>&lt;.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>7</td>
<td>3</td>
<td>N</td>
<td>7</td>
<td>12</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
</tr>
<tr>
<td>78</td>
<td>ACM-511</td>
<td>67-A 51-511</td>
<td>&lt;.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>7</td>
<td>3</td>
<td>N</td>
<td>7</td>
<td>12</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
</tr>
<tr>
<td>79</td>
<td>ACM-513</td>
<td>67-A 51-513</td>
<td>&lt;.02</td>
<td>N</td>
<td>ND</td>
<td>N</td>
<td>7</td>
<td>3</td>
<td>N</td>
<td>7</td>
<td>12</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
<td>N</td>
</tr>
</tbody>
</table>

1 Where duplicate values are reported, highest value was used to compute anomaly.
Figure 2.—Distribution of metals in panned concentrates and stream sediments, Humboldt Creek. Panned concentrates obtained from all localities; underscoring indicates that stream-sediment sample was also obtained.
EXPLANATION

SAMPLE LOCALITY AND NUMBER

\(\bullet^1\)
No anomalous metals

\(\bullet^2 \circ^37 \circ^44\)
1-3 metals anomalous in amount

\(\circ^34\)
4-7 metals anomalous in amount

\(\bullet^{15}\)
More than 10 metals anomalous in amount

Size of symbol indicates sum of anomalous metals, in parts per million:

Small, in amount between 1 and 5 times background
Medium, in amount between 6 and 25 times background
Large, in amount greater than 25 times background
Of all the metals, mercury shows the most clear-cut, direct correlation with the total metal anomaly, although it seldom accounts for a large share of the total anomaly.

A marked inverse relationship exists between manganese and total trace metals in the altered zones sampled. Whether this relationship represents a selective leaching of manganese during supergene alteration of sulfides, or whether it represents leaching of manganese by ore solutions, is not known. In either case, the absence of manganese from mineralized samples is a direct indication of possible mineralized structures. Lyman Huff (oral commun., 1968) reported an inverse relation between manganese and copper near certain porphyry copper deposits he is currently studying.

SUGGESTIONS FOR EXPLORATION

Future work in the Serpentine Hot Springs area should include the following:

1. Detailed mapping and sampling of all observable alteration areas, including those along the thrust faults.

2. Determination of the bedrock source of metals leading to the geochemical anomalies in the west fork of Humboldt Creek (sample locs. 43, 44, fig. 2), in the west headwaters of the Pish River, and in the bedrock east of Humboldt Creek.

3. Trenching, and possibly drilling, of the two main mineralized zones, now known, or others that might be proved by mapping and sampling.

4. Evaluation, by churn drilling, of the tin and gold potential of placer deposits in the entire Humboldt Creek drainage, as well as in the upper reaches of Kennedy Creek, which heads against the mineralized area shown by sample localities 56–61 (pl. 1), and in the west fork of the Pish River.

5. Careful mineralogical work on samples of argentiferous galena to determine if the high silver values are caused by secondary enrichment, which, in this area, would probably occur only near the surface. If high silver values persist to depth, lode mining may be profitable even in this isolated part of Alaska.

REFERENCES CITED


### Table 4.—Metal content, in parts per million, of stream sediments and panned concentrates of stream sediments, Humboldt Creek drainage, Serpentine Hot Springs area, Alaska

<table>
<thead>
<tr>
<th>Field No.</th>
<th>Sample type</th>
<th>Sample type</th>
<th>As</th>
<th>Ag</th>
<th>Au</th>
<th>Bi</th>
<th>Cd</th>
<th>Co</th>
<th>Cu</th>
<th>Mo</th>
<th>Pb</th>
<th>Sn</th>
<th>Zn</th>
</tr>
</thead>
<tbody>
<tr>
<td>67-ASn-543</td>
<td>S, -80 mesh</td>
<td>&lt;0.02</td>
<td>67-ASn-543</td>
<td>S, -80 mesh</td>
<td>&lt;0.02</td>
<td>67-ASn-543</td>
<td>S, -80 mesh</td>
<td>&lt;0.02</td>
<td>67-ASn-543</td>
<td>S, -80 mesh</td>
<td>&lt;0.02</td>
<td>67-ASn-543</td>
<td>S, -80 mesh</td>
</tr>
<tr>
<td>67-AKd-145</td>
<td>S, -80 mesh</td>
<td>&lt;0.02</td>
<td>67-AKd-145</td>
<td>S, -80 mesh</td>
<td>&lt;0.02</td>
<td>67-AKd-145</td>
<td>S, -80 mesh</td>
<td>&lt;0.02</td>
<td>67-AKd-145</td>
<td>S, -80 mesh</td>
<td>&lt;0.02</td>
<td>67-AKd-145</td>
<td>S, -80 mesh</td>
</tr>
<tr>
<td>67-AKd-39</td>
<td>S, -40 mesh</td>
<td>&lt;0.02</td>
<td>67-AKd-39</td>
<td>S, -40 mesh</td>
<td>&lt;0.02</td>
<td>67-AKd-39</td>
<td>S, -40 mesh</td>
<td>&lt;0.02</td>
<td>67-AKd-39</td>
<td>S, -40 mesh</td>
<td>&lt;0.02</td>
<td>67-AKd-39</td>
<td>S, -40 mesh</td>
</tr>
<tr>
<td>68-ATr-65</td>
<td>PC</td>
<td>&lt;0.02</td>
<td>68-ATr-65</td>
<td>PC</td>
<td>&lt;0.02</td>
<td>68-ATr-65</td>
<td>PC</td>
<td>&lt;0.02</td>
<td>68-ATr-65</td>
<td>PC</td>
<td>&lt;0.02</td>
<td>68-ATr-65</td>
<td>PC</td>
</tr>
<tr>
<td>66-ATr-65</td>
<td>PC</td>
<td>&lt;0.02</td>
<td>66-ATr-65</td>
<td>PC</td>
<td>&lt;0.02</td>
<td>66-ATr-65</td>
<td>PC</td>
<td>&lt;0.02</td>
<td>66-ATr-65</td>
<td>PC</td>
<td>&lt;0.02</td>
<td>66-ATr-65</td>
<td>PC</td>
</tr>
<tr>
<td>66-AKd-65</td>
<td>S, -40 mesh</td>
<td>&lt;0.02</td>
<td>66-AKd-65</td>
<td>S, -40 mesh</td>
<td>&lt;0.02</td>
<td>66-AKd-65</td>
<td>S, -40 mesh</td>
<td>&lt;0.02</td>
<td>66-AKd-65</td>
<td>S, -40 mesh</td>
<td>&lt;0.02</td>
<td>66-AKd-65</td>
<td>S, -40 mesh</td>
</tr>
<tr>
<td>67-ASn-543</td>
<td>S, -80 mesh</td>
<td>&lt;0.02</td>
<td>67-ASn-543</td>
<td>S, -80 mesh</td>
<td>&lt;0.02</td>
<td>67-ASn-543</td>
<td>S, -80 mesh</td>
<td>&lt;0.02</td>
<td>67-ASn-543</td>
<td>S, -80 mesh</td>
<td>&lt;0.02</td>
<td>67-ASn-543</td>
<td>S, -80 mesh</td>
</tr>
<tr>
<td>67-AKd-145</td>
<td>S, -80 mesh</td>
<td>&lt;0.02</td>
<td>67-AKd-145</td>
<td>S, -80 mesh</td>
<td>&lt;0.02</td>
<td>67-AKd-145</td>
<td>S, -80 mesh</td>
<td>&lt;0.02</td>
<td>67-AKd-145</td>
<td>S, -80 mesh</td>
<td>&lt;0.02</td>
<td>67-AKd-145</td>
<td>S, -80 mesh</td>
</tr>
</tbody>
</table>

*Note: Double entry indicates duplicate determinations; highest value is used to compute anomaly. Analyses: R. L. Miller, D. J. Atwood, and D. E. Martinez. Precision: numbers below detection limit, in descending order, 2 ppm, 1 ppm, and detection limit (ND) in bozeheds, numbers beneath each analytical result type. Metals present in anomalous amounts and their approximate concentration limits: Au, 5 ppm; Ag, 2 ppm; Cu, 1 ppm; Ni, 0.5 ppm; Zn, 10 ppm; Mo, 0.5 ppm; Pb, 1 ppm; Sn, 0.5 ppm; Co, 0.5 ppm. All analyses are semiquantitative spectrophotometric except those for Au, which are wet chemical. Double entry indicates duplicate determinations; highest value is used to compute anomaly. Analysts: R. L. Miller, D. J. Atwood, and D. E. Martinez."